

Mehmet Akkurt,^a Sema Öztürk,^{a*} Tefvik Rıza Kök^b and Hoong-Kun Fun^c^aDepartment of Physics, Faculty of Arts and Sciences, Erciyes University, 38039 Kayseri, Turkey, ^bDepartment of Chemistry, Faculty of Arts and Sciences, Erciyes University, 38039 Kayseri, Turkey, and ^cX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

Correspondence e-mail: ozturk@erciyes.edu.tr

Key indicators

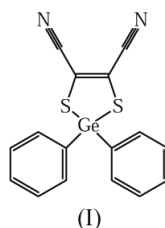
Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.007$ Å
 R factor = 0.045
 wR factor = 0.100
Data-to-parameter ratio = 21.0For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

2,2-Diphenyl-1,3,2-dithiagermole-4,5-dicarbonitrile

The title compound, $(\text{C}_6\text{H}_5)_2\text{GeS}_2\text{C}_2(\text{CN})_2$, crystallizes with two molecules in the asymmetric unit. Each five-membered ring has an envelope conformation.Received 11 July 2003
Accepted 23 July 2003
Online 31 July 2003

Comment

The X-ray investigation of the title compound, (I), was undertaken as part of our study of the structure and conformation of new germanium complexes.

In the title compound, (I), the average Ge–S and Ge–C bond lengths are 2.2530 (13) and 1.926 (5) Å, respectively, in agreement with the literature values (Allen *et al.*, 1987; Liu *et al.*, 2002).The title compound crystallizes with two molecules (denoted *A* and *B*) in the asymmetric unit. In molecule *A*, the dihedral angle between the five-membered ring and phenyl ring C1A–C6A is 82.4 (1)°, that between the five-membered ring and phenyl ring C7A–C12A is 66.1 (1)° and the angle between the two phenyl rings is 55.8 (2)°. The corresponding angles in molecule *B* are 73.1 (1), 79.1 (1) and 58.9 (1)°, respectively. Each five-membered ring has an envelope conformation, the flaps Ge1A and Ge1B being displaced from the planes of the other ring atoms by 0.4115 (4) and 0.3467 (5) Å, respectively. In 2,2-dimethyl-[1,3,2]dithiagermole-4,5-dicarbonitrile, $\text{Me}_2\text{GeS}_2\text{C}_2(\text{CN})_2$ (Ewert, 1983), the five-membered ring also has an envelope conformation.

Experimental

Under an argon atmosphere, a stirred aqueous solution of $\text{Na}_2\text{S}_2\text{C}_2(\text{CN})_2$ was added dropwise to a solution of Ph_2GeBr_2 in CH_3Cl (1:1). This mixture was stirred at room temperature for 1 h. After filtration, the coloured solid was dried over anhydrous MgSO_4 . The solvent was then evaporated and the product crystallized from absolute toluene. Analysis calculated for $\text{C}_{16}\text{H}_{10}\text{GeN}_2\text{S}_2$: C 52.10, H 2.82, Ge 19.00, S 17.39%; found: C 52.32, H 2.74, Ge, 19.78, S 17.47%. M.p.: 416 K. IR, ν (cm^{-1}): 2220 (C–N), 700 (C–S), 405 (Ge–S), 310 (Ge–C). ^1H NMR (δ , p.p.m., 200 MHz): 7.56. ^{13}C NMR (δ , p.p.m., 200 MHz): 132.25 (α), 133.84 (β), 129.60 (γ), 132.44 (δ), 119.63 (C=C), 112.96 (CN).

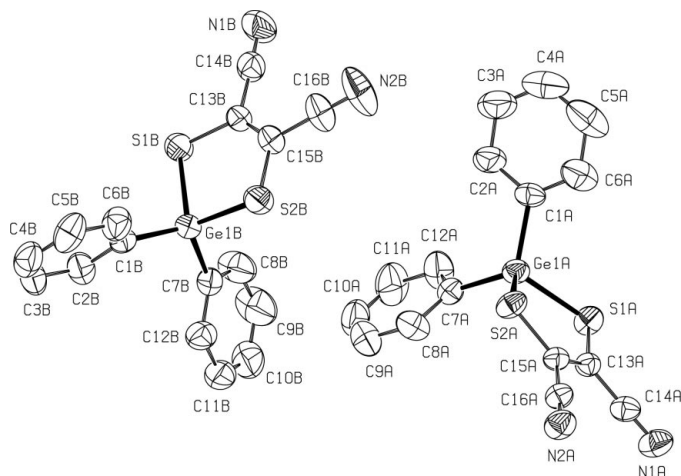


Figure 1
Views of the two independent molecules of (I), with the atom-numbering scheme and displacement ellipsoids drawn at the 50% probability level. H atoms have been omitted.

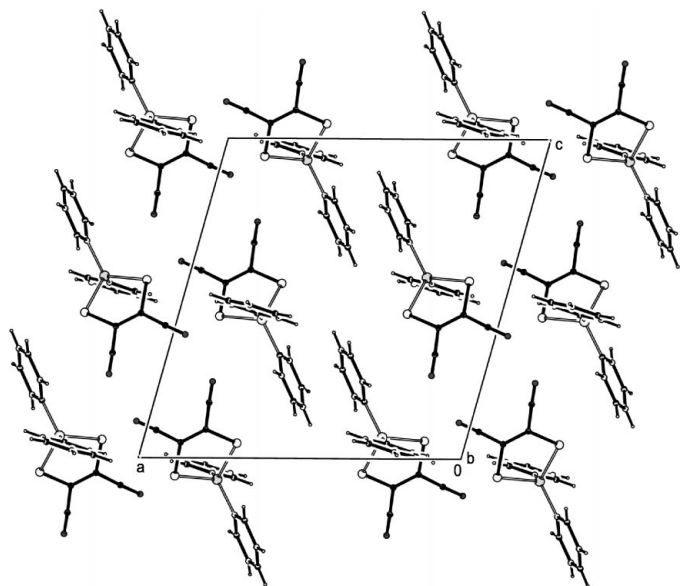


Figure 2
A packing diagram of the title compound, (I), viewed along the *b* axis.

Crystal data

$C_{16}H_{10}GeN_2S_2$
 $M_r = 367.01$
 Monoclinic, $P2_1/c$
 $a = 15.3121 (11) \text{ \AA}$
 $b = 14.0064 (10) \text{ \AA}$
 $c = 15.7730 (11) \text{ \AA}$
 $\beta = 105.3900 (10)^\circ$
 $V = 3261.5 (4) \text{ \AA}^3$
 $Z = 8$

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.483$, $T_{\max} = 0.676$
 19 939 measured reflections

$D_x = 1.495 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation
 Cell parameters from 5356 reflections
 $\theta = 2.4\text{--}24.7^\circ$
 $\mu = 2.13 \text{ mm}^{-1}$
 $T = 293 \text{ K}$
 Slab, light brown
 $0.40 \times 0.40 \times 0.20 \text{ mm}$

7949 independent reflections
 4421 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$
 $\theta_{\text{max}} = 28.3^\circ$
 $h = -16 \rightarrow 20$
 $k = -18 \rightarrow 13$
 $l = -20 \rightarrow 21$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.100$
 $S = 1.04$
 7949 reflections
 379 parameters
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0286P)^2 + 1.6083P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.39 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.34 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

Ge1A—S1A	2.2484 (11)	S1A—C13A	1.742 (3)
Ge1A—S2A	2.2616 (12)	S2A—C15A	1.751 (3)
Ge1A—C1A	1.926 (3)	S1B—C13B	1.737 (4)
Ge1A—C7A	1.931 (4)	S2B—C15B	1.742 (5)
Ge1B—C1B	1.922 (3)	N1A—C14A	1.136 (6)
Ge1B—C7B	1.926 (5)	N2A—C16A	1.135 (6)
Ge1B—S2B	2.2480 (11)	N1B—C14B	1.123 (7)
Ge1B—S1B	2.2539 (13)	N2B—C16B	1.136 (6)
S1A—Ge1A—S2A	95.16 (4)	Ge1A—C7A—C8A	121.6 (3)
S1A—Ge1A—C1A	112.76 (11)	Ge1A—C7A—C12A	121.4 (3)
S1A—Ge1A—C7A	113.55 (11)	S1A—C13A—C14A	114.1 (3)
S2A—Ge1A—C1A	110.94 (11)	S1A—C13A—C15A	124.8 (2)
S2A—Ge1A—C7A	108.90 (14)	N1A—C14A—C13A	178.2 (5)
C1A—Ge1A—C7A	113.94 (16)	S2A—C15A—C16A	114.5 (2)
S1B—Ge1B—S2B	95.40 (4)	S2A—C15A—C13A	124.3 (2)
S1B—Ge1B—C1B	109.02 (11)	N2A—C16A—C15A	177.6 (4)
S1B—Ge1B—C7B	109.22 (14)	Ge1B—C1B—C2B	120.0 (3)
S2B—Ge1B—C1B	114.03 (12)	Ge1B—C1B—C6B	121.9 (3)
S2B—Ge1B—C7B	111.25 (13)	Ge1B—C7B—C8B	119.6 (3)
C1B—Ge1B—C7B	115.88 (17)	Ge1B—C7B—C12B	121.4 (4)
Ge1A—S1A—C13A	96.94 (11)	S1B—C13B—C14B	114.0 (3)
Ge1A—S2A—C15A	96.45 (11)	S1B—C13B—C15B	124.8 (3)
Ge1B—S1B—C13B	96.71 (12)	N1B—C14B—C13B	178.2 (5)
Ge1B—S2B—C15B	96.78 (15)	S2B—C15B—C13B	124.6 (3)
Ge1A—C1A—C2A	119.4 (3)	S2B—C15B—C16B	115.3 (3)
Ge1A—C1A—C6A	121.2 (3)	N2B—C16B—C15B	179.3 (5)
Ge1A—S1A—C13A—C14A	171.5 (2)	Ge1B—S1B—C13B—C15B	−8.9 (3)
Ge1A—S1A—C13A—C15A	−9.3 (3)	Ge1B—S1B—C13B—C14B	174.4 (3)
Ge1A—S2A—C15A—C13A	11.8 (3)	Ge1B—S2B—C15B—C13B	9.0 (4)
Ge1A—S2A—C15A—C16A—171.7 (2)		Ge1B—S2B—C15B—C16B	−170.5 (3)

All H atoms were placed in geometrically idealized positions, with C—H distances of 0.93 \AA , and refined in the riding-model approximation, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}$ of their parent atoms.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
 Ewert, I. (1983). MSc thesis, University of Bonn, Germany.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
 Liu, Y., Ballweg, D., Müller, T., Guzei, I. A., Clark, R. W. & West, R. (2002). *J. Am. Chem. Soc.* **124**, 12174–12181.
 Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
 Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.